

Solutions Chapter 2



Solutions Chapter 2

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2.10 What role does the molecular interaction play in a solution of alcohol and water? Sol. Alcohol and water both have strong tendency to form intermolecular hydrogen bonding. On mixing the two, a solution is formed as a result of formation of H-bonds between alcohol and H₂O molecules but these interactions are weaker and less extensive than those in pure H₂O.

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Chemistry Notes for class 12 Chapter 2 Solutions Solution is a homogeneous mixture of two or more substances in same or different physical phases. The substances forming the solution are called components of the solution. On the basis of number of components a solution of two components is called binary solution. Solute and Solvent

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CHAPTER 2: CONSOLIDATION OF FINANCIAL INFORMATION

Solutions for Section 2.2 Exercise 2.2.1(a) States correspond to the eight combinations of switch positions, and also must indicate whether the previous roll came out at D, i.e., whether the previous input was accepted. Let 0 represent a position to the left (as in the diagram) and 1 a position to the right.

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